

I'm not a robot



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1 2 3 4 5 6 trials.

Data table for acceleration of a lab cart on a horizontal surface.

$$m_c + m_w = 0.3381 \text{ kg}$$

m_c	m_w	Applied Force to the system = $m_w g$	$a_{\text{theoretical}}$	a_{measured}	percent difference
0.3203 kg	0.0178 kg	0.17444 N	0.5313 m/s^2	0.505 m/s^2	5.08%
0.3123 kg	0.0258 kg	0.25284 N	0.7701 m/s^2	0.7108 m/s^2	8.01%
0.3043 kg	0.0338 kg	0.33124 N	1.009 m/s^2	0.9282 m/s^2	8.34%
0.2963 kg	0.0418 kg	0.40964 N	1.2478 m/s^2	1.156 m/s^2	7.64%
0.2883 kg	0.0498 kg	0.48804 N	1.4866 m/s^2	1.363 m/s^2	8.67%

1. Construct a graph of Force vs. acceleration. (y vs. x) in Graphical Analysis. Include the graph and relevant analysis with your lab report.
(SEE ATTACHED GRAPH)
2. Interpret the slope and y-intercept (if applicable) using your pre-lab work.
3. Include a sample error calculation. What are possible sources of error?
4. State briefly and precisely - what did you explore in this portion of the lab and what did you find and learn.

Part 2: Inclined plane.

Theory 2: A light string connects a cart and weight. The weight falls under the influence of gravity, pulling the cart over a frictionless pulley. Now the cart is moving on an inclined plane.



DOCHERO

Source for or Credit None or pg. 45 Topic	12. How Sound Travels Through the Inner Ear
12. How Sound Travels Through the Inner Ear	Score _____
Essential Question x None	

Lab Report Worksheet: What Are the Kinetics of an Iodine Clock?

Lab Section or Day-Time Your Name: _____

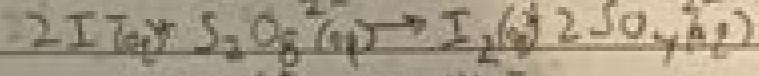
Your Team: Team 6

This worksheet does not replace good lab notebook keeping!

ALL INFORMATION RECORDED ON THIS WORKSHEET

**ALL INFORMATION REQUESTED ON THIS FORM
SHOULD COME FROM THE BOOK, PRELAB NOTES, THE LAB MANUAL AND
PROCEDURES/VALUES/CALCULATIONS YOU HAVE RECORDED IN YOUR NOTEBOOK!**

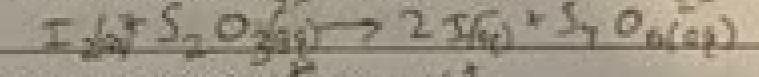
1. What is the balanced net ionic equation for the main reaction between KI and $(\text{NH}_4)_2\text{S}_2\text{O}_8$?



What is/are the spectator ion(s)? K^+ AND NH_4^+

Is this reaction fast or slow? slow

2. What is the net ionic equation for the clock reaction?



What is/are the spectator ion(s)? NO_3^- AND K^+

Is this reaction fast or slow? Fast

What reagent is added that will change color? What is it reacting with? What color will it turn?

JANUARY 100.00 DAILY VALUE

Which reactant must be used up before the indicator changes color? $\text{Na}_2\text{S}_2\text{O}_3$

: Determination of Rate Law

What is the generic form of the

You are given suggested volumes of reagents for performing

main reagents (the I⁻ or the S₂O₈²⁻) to discover the effect of the concent

Which of the main reagents was in the beaker? I

Which of the main reagents was in the test tube? $S_2 O_8^{2-}$

Which reagent was added to keep a constant volume and constant concentration of K^+ ? Li_2O_2

Which reagent was added to keep a constant volume and constant concentration of NH_4^+ ? (Nessler's, 2

FILL IN THE EXPERIMENTAL VOLUMES (IN DROPS) FOR RUNS 1-5 INTO TABLE

Table 1: Experimental Volumes for Iodine Clock Reactions for Part 1

Table 1: Experimental volumes for yeast extract dilutions for FRT101

	Initial Delever	Test Date				
	1%	0.01M	0.20M	0.20M	0.20M	0.20M

starch $\text{Na}_2\text{S}_2\text{O}_3$ KI KNO_3 $(\text{NH}_4)_2\text{S}_2\text{O}_3$

	Sample	(Nanomolar)	RI	RMSD	(Nanometer)	(Nanometer)
Run 1	3	5	20	0	10	10
Run 2	3	5	10	10	10	10
Run 3	3	5	5	15	10	10
Run 4	3	5	10	10	20	0
Run 5	3	5	10	10	5	15

<http://www.ncbi.nlm.nih.gov> | <http://www.ncbi.nlm.nih.gov/entrez> | <http://www.ncbi.nlm.nih.gov/blast>

Lab 31 smell and taste. Laboratory report 31 smell and taste

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